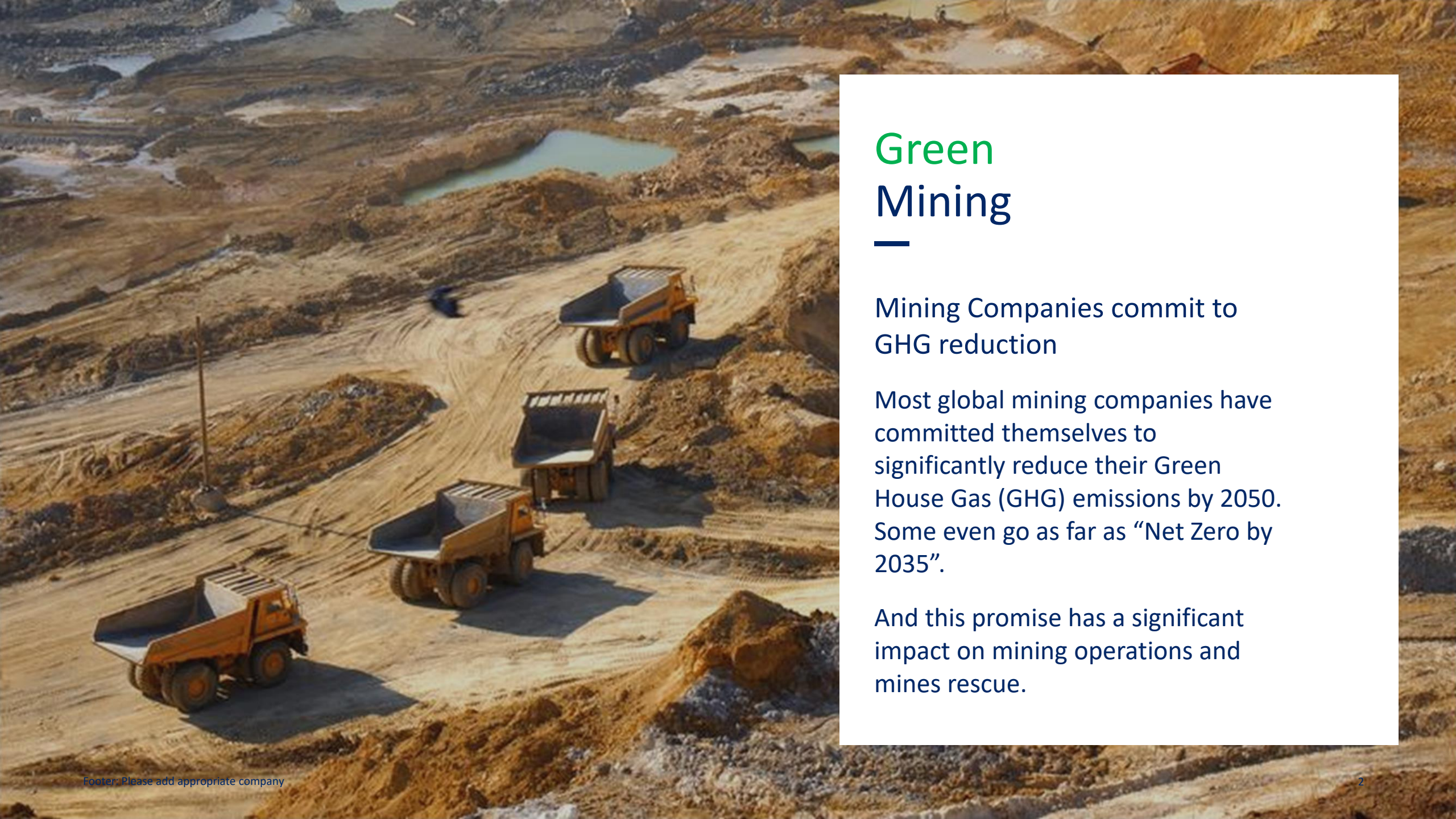


Every Coin has 2 Sides: Green Energy and Mines Rescue

Presentation by Axel Bahr, Dräger Safety





Green Mining

Mining Companies commit to
GHG reduction

Most global mining companies have committed themselves to significantly reduce their Green House Gas (GHG) emissions by 2050. Some even go as far as “Net Zero by 2035”.

And this promise has a significant impact on mining operations and mines rescue.

Clean Energy in the Mining Industry

“There are five focus decarbonisation technologies in the mining industry, and a range of energy transition technologies will need to be deployed to decarbonise mining operations” (Alex Phillips, energy transition analyst at GlobalData)

1. Electrification

of mining equipment, especially haulers and diggers

2. Hydrogen

for use in Fuel Cells or direct combustion (e.g. smelters)

3. Renewable Energy

from Solar and Wind farms

4. Alternative Fuels

such as Green Methanol for combustion engines

5. Carbon Capture, Utilisation and Storage

to reduce CO₂ release

Clean Energy Safety Solutions

Item 01

What is Clean Energy and
what are the risks?



Item 02

What is Dräger's safety
approach for Clean
Energy?



Item 03
Hydrogen

Item 04
Recommendation



01 What is Clean Energy and what are the risks?

Clean Energy Hazards

Hydrogen (H₂)

- Highly flammable
- Invisible flame
- Odorless / colorless
- High risk of leakages

Batteries (HF, CO, CO₂, HCN)

- Highly toxic
- Highly flammable
- Frostbite



CCUS (CO₂)

- Very toxic
- Strong green house gas

Methanol (CH₃OH)

- Very toxic
- Highly flammable
- Almost invisible flame

Clean Energy

Hydrogen (H₂) Hazards



Explosion

A H₂-air-mixture is highly flammable in a wide range between 4%Vol and 77%Vol. The energy needed to ignite this mixture is 0,017mJ which further increases the risk of explosions or even self-ignition.



Invisible Flame

In daylight H₂ flames are nearly invisible. In addition they hardly radiate heat and are odorless making them extremely difficult to notice.



Leakages

Due to its small molecules and low viscosity, H₂ is prone to leak from seals and valves. High pressure leaks pose a significant risk of self-ignition

Clean Energy

Batteries (HF, CO, CO₂, HCN) Hazards



Explosion

Battery-cell thermal runaway leads to high risk of explosion where gases and vapors can accumulate in poorly ventilated areas



Flammability

During incidents, clouds of flammable gases and vapors are released from the cells. These flammable vapor clouds can be sometimes mistaken for smoke or steam.



Toxicity

Toxic and irritant emissions during thermal runaway or fires (e.g. CO, HF, HCl, HCN, SO₂, NO₂, and PAHs)

Clean Energy

Methanol (CH_3OH) Hazards



Toxicity

Methanol is very toxic and exposure can damage specific organs and the central nervous system.



Flammability

With a lower explosion limit of 6%Vol. an upper limit of 50%Vol, a flash point of 9°C and an ignition temperature of just 440°C, Methanol is considered to be a highly flammable substance.

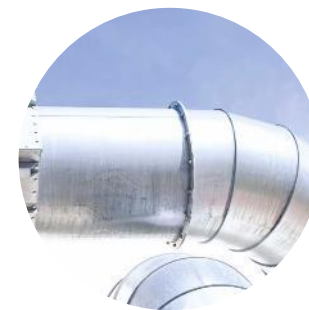


Invisible Flame

The flame of a Methanol fire is very pale and nearly invisible in daylight. Unlike Hydrogen, the flames emit radiant heat and can be felt from a distance.

Clean Energy

CCUS (CO₂) Hazards



Toxicity

Breathing in high concentrations of CO₂ leads to intoxication resulting in sleepiness, loss of concentration and ultimately respiratory arrest.



Greenhouse Gas

CO₂ is the most significant long-lived greenhouse gas in earth's atmosphere.

The rapid increase in its concentration lead to global warming.



02

What is Dräger's safety
approach for **Clean
Energy**

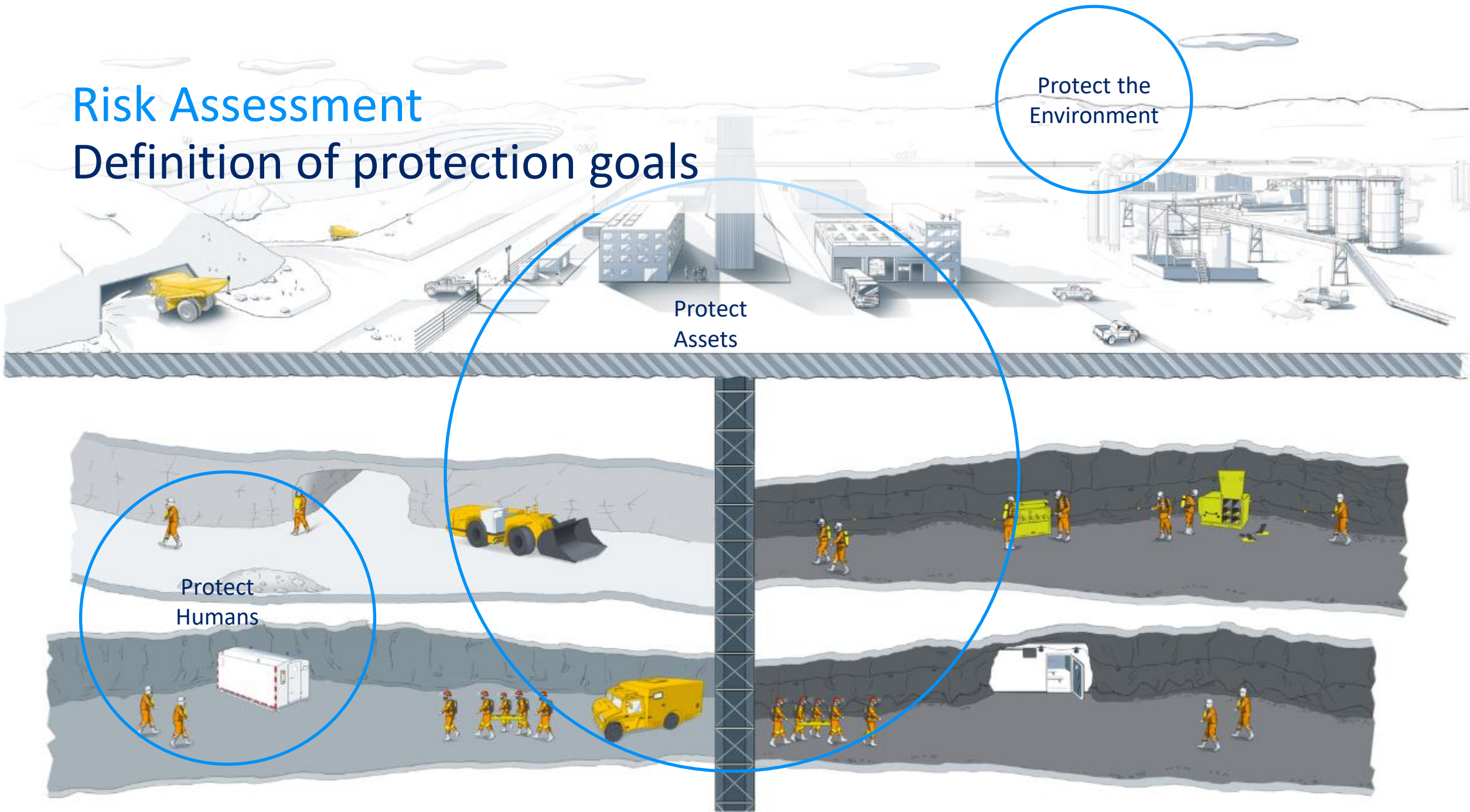
Risk Assessment

Definition of protection goals

Protect the
Environment

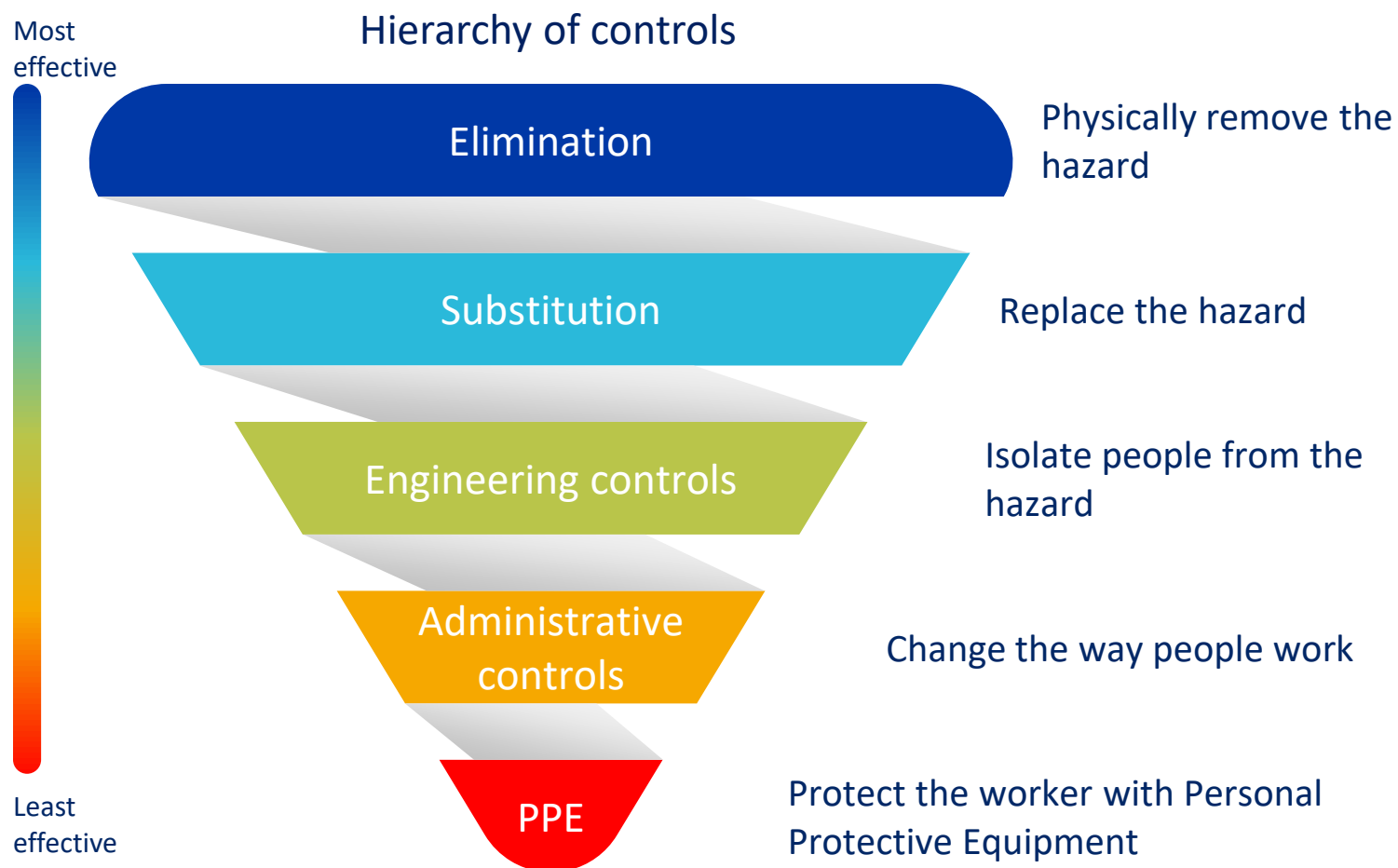
Protect
Assets

Protect
Humans



Risk assessment

Selection of protective measures



Fire and Gas Detection System

- Explosion-proof electrical equipment
- Technical safety (e.g. TPRD)
- Ventilation equipment



Breathing protection

- CCBA
- SCBA
- Masks & Filters





03 Hydrogen

Safety Challenges

Hydrogen



Education
Threats from H₂ should neither be overrated, nor underestimated.



Permeation
H₂ is the smallest of all molecules and can permeate through and embrittle materials. The right choice, handling and maintenance of materials is critical.



Leaks
Due to its small molecular size and low viscosity, H₂ can quickly leak from its pipelines.



Gas pockets
H₂ is less dense than air and forms gas pockets at indoor ceilings.



Odorless
Not only is H₂ colorless, it is also odorless. Therefore, a person cannot detect hydrogen.



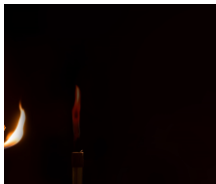
Ex-protection
The biggest threat from hydrogen is that it leads to an explosion. Equipment in Ex-areas must be suitable and certified.



Invisible for IR
IR sensors in gas detection devices cannot detect H₂ because of missing C-H bonds.



Invisible Flame
The hydrogen flame is very pale and not visible in daylight.



DANGER

HYDROGEN IN USE



NOT VISIBLE



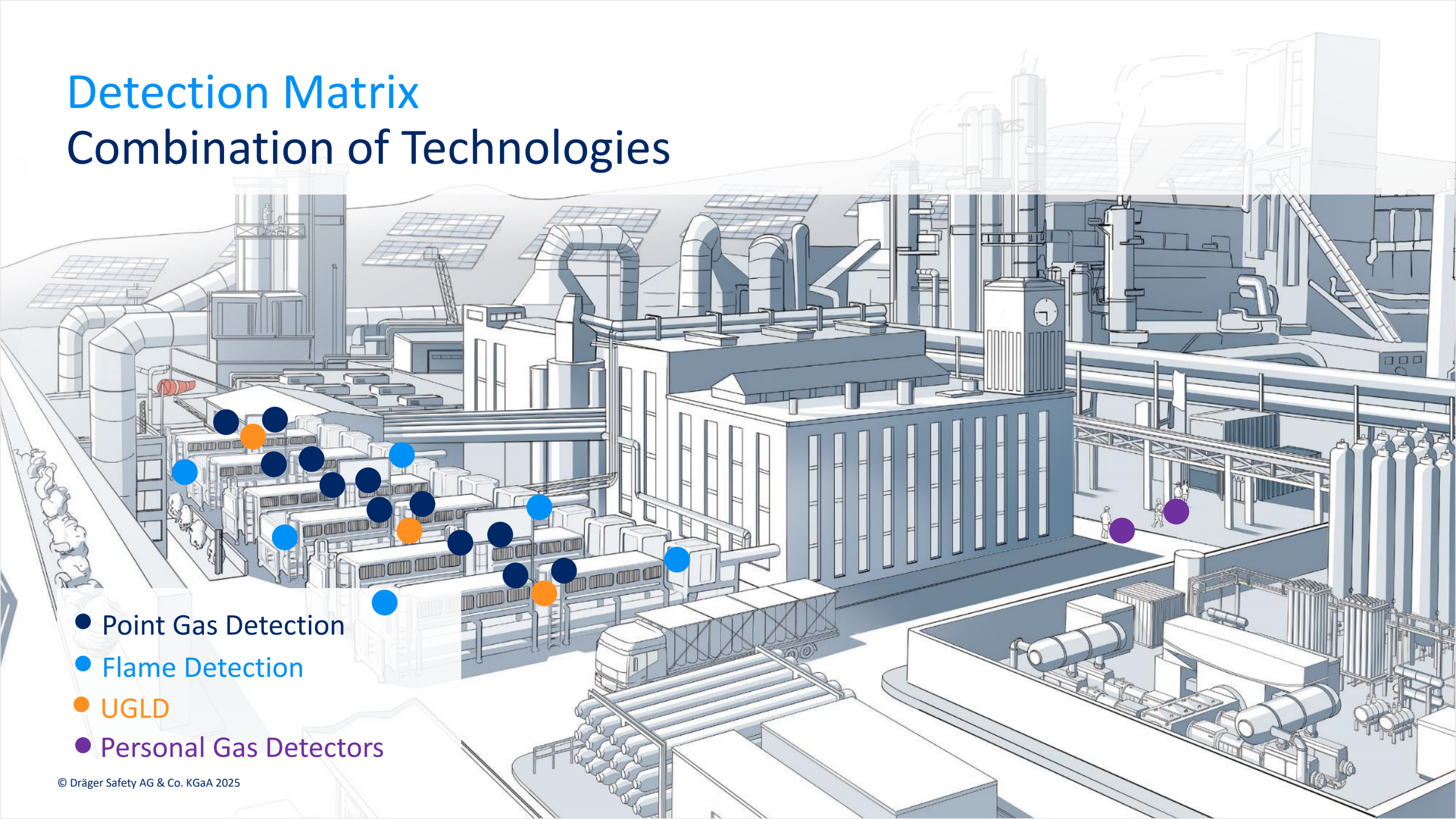
NO ODOR

Dräger's Detection Matrix approach

In order to ensure the fast and reliable detection of Hydrogen, Ammonia, Methanol or Carbon Dioxide, we recommend a matrix of different detection technologies to combine each technology's strength and eradicate its weakness.

Detection Matrix

Combination of Technologies

- 
- The background image is a detailed line drawing of an industrial plant. It features various structures including storage tanks, distillation columns, piping, and a central building with a clock tower. In the foreground, there are stacks of pipes and a truck. Colored dots are placed throughout the scene to indicate the locations of different detection technologies: dark blue dots for point gas detection, light blue dots for flame detection, orange dots for UGLD, and purple dots for personal gas detectors. The dots are concentrated in the central and left areas of the plant, with a few purple dots near the clock tower building.
- Point Gas Detection
 - Flame Detection
 - UGLD
 - Personal Gas Detectors

Fire and Gas Detection System

Senses vs. Sensor Technologies

Point Detector



Flammable IR Gas Detector



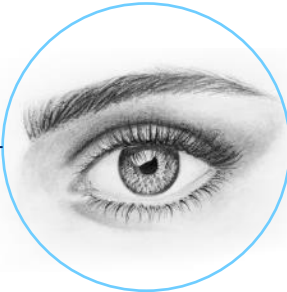
Toxic Electrochemical Gas Detector



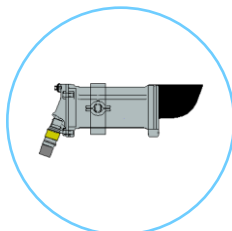
Flammable Catalytic Gas Detector



Area Monitoring / Line of sight



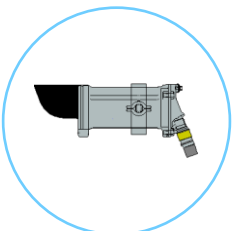
Flame Detector



Open Path Gas Detector



Gas Camera



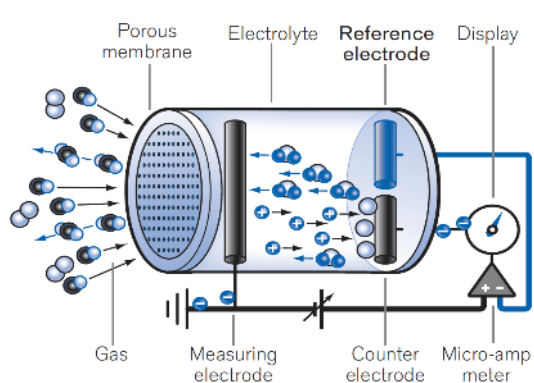
Ultrasonic Gas Leak Detector



UGLD

Basic gas detection technologies

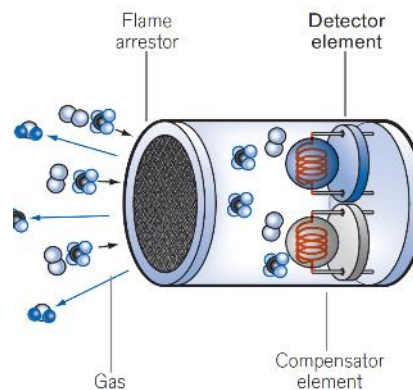
Point gas detectors



Electrochemical sensor (EC)

EC sensors function **similar to batteries** and detect gases in **ppm concentrations**.

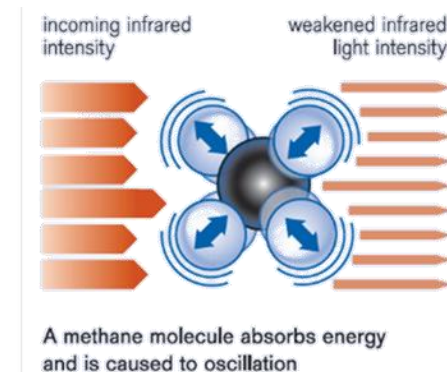
- Target substance **reacts chemically** and generates an **electrical charge** between two electrodes.
- The signal strength is proportional to the gas concentration.



Catalytic bead sensor (CatEx)

CatEx sensors **oxidize combustible** gases, releasing heat.

- A **suitable temperature-controlled catalyst** material is used, which is additionally heated measurably by the reaction heat.
- This small increase in temperature is a measure of the gas concentration.



Infrared Sensor (IR)

IR sensors measure the **change in intensity** of infrared radiation absorbed by target materials.

- Infrared is weakened by C-H, but not by H-H bonds.
- Methane can be measured, **but hydrogen cannot**.

Portable Gas Detection

H₂ Sensor Technologies

NEW



Dräger Pac 8000 H₂

Personal Air Monitoring and Protection

- DrägerSensor XXS H₂ HC:
0 – 4 Vol% / 0 - 100 % LEL



Dräger X-am 5800

Personal Air Monitoring and Protection

- Ex-protection by CatEx sensor
up to 100 % LEL
- DrägerSensor XXS H₂ HC:
0 – 4 Vol% / 0 - 100 % LEL



Dräger X-am 8000

Leak Detection and Clearance Measurements

- Ex-protection by CatEx sensor optimized for H₂ with 3 measurements ranges: up to 40,000 ppm, 100 % LEL and 0 - 100 Vol.-% (NEW)
- DrägerSensor XXS H₂: 0 - 2,000 ppm
- DrägerSensor XXS H₂ HC:
0 - 40,000 ppm / 0 - 100 % LEL

Fire and Gas Detection System

Senses vs. Sensor Technologies

Point Detector



Flammable IR Gas Detector



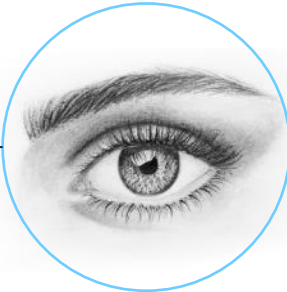
Toxic Electrochemical Gas Detector



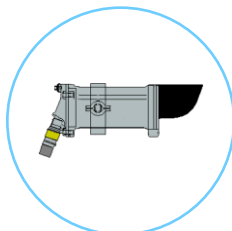
Flammable Catalytic Gas Detector



Area Monitoring / Line of sight



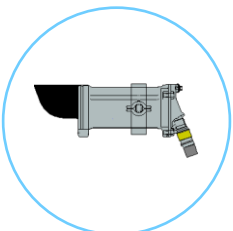
Flame Detector



Open Path Gas Detector



Gas Camera



Ultrasonic Gas Leak Detector



UGLD

Fire and Gas Detection System

Hydrogen Flame Detection



To the human body and all our senses hydrogen and its flames are almost undetectable.

Hydrogen flames

- are invisible in daylight
- emit very little radiant heat
- are odorless



In addition, Hydrogen has the following unique characteristics that make it more hazardous than other substances:

- LEL at 20°C is 4 Vol%
- UEL at 20°C is 77 Vol%
- Ignition energy is 0,017 mJ

Fire and Gas Detection System

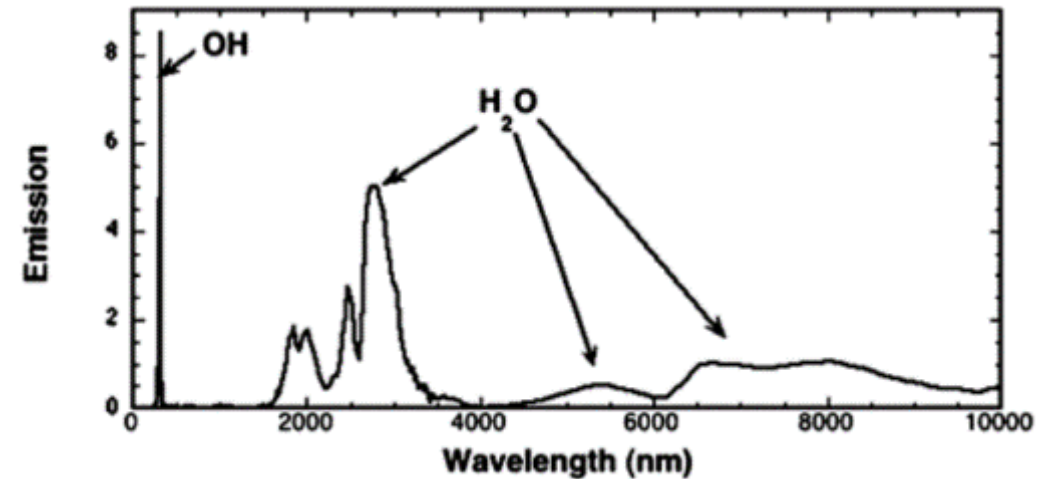
Hydrogen Flame Detection



Hydrogen flames do **NOT** emit CO₂



Hydrogen flames only emit hot H₂O

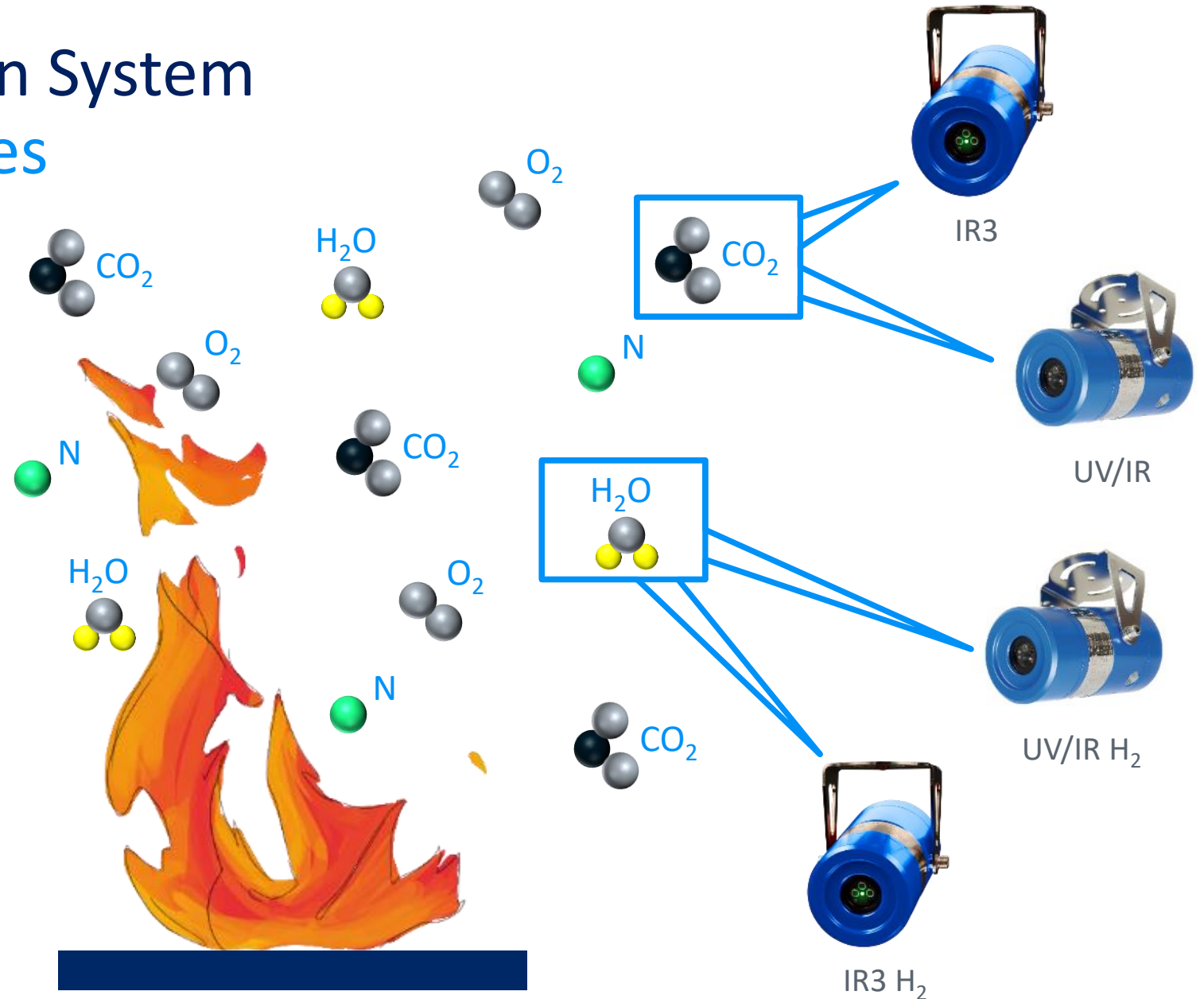


Fire and Gas Detection System Detection Technologies



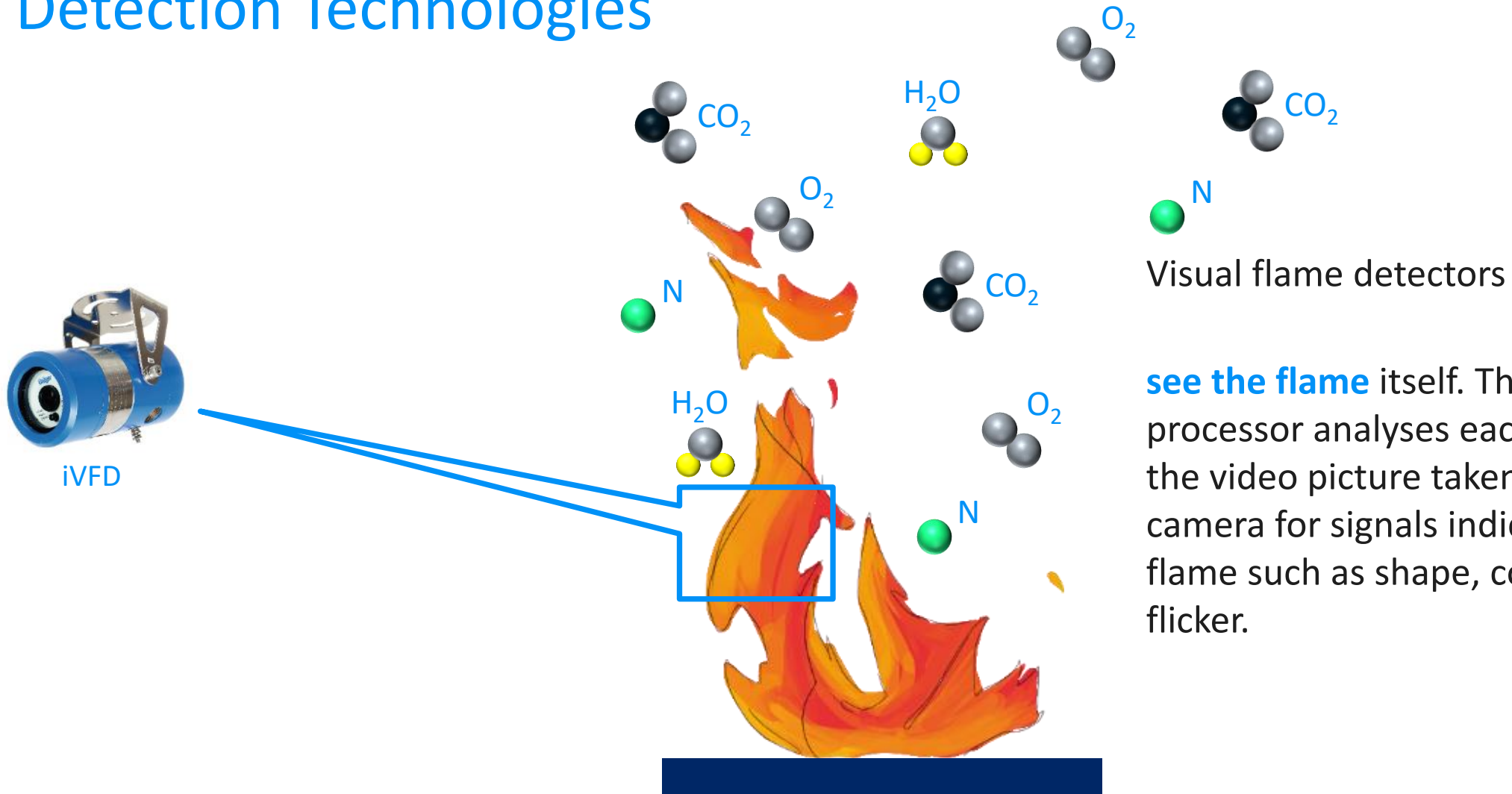
Radiant flame detectors

do NOT see the flame itself. They only see the electromagnetic radiation emitted by the combustion products. Their frequency spectrum depends on the chemical composition of the reaction products.



Fire and Gas Detection System

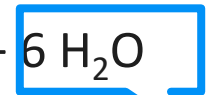
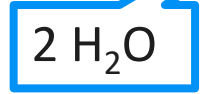
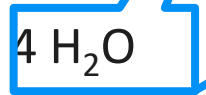
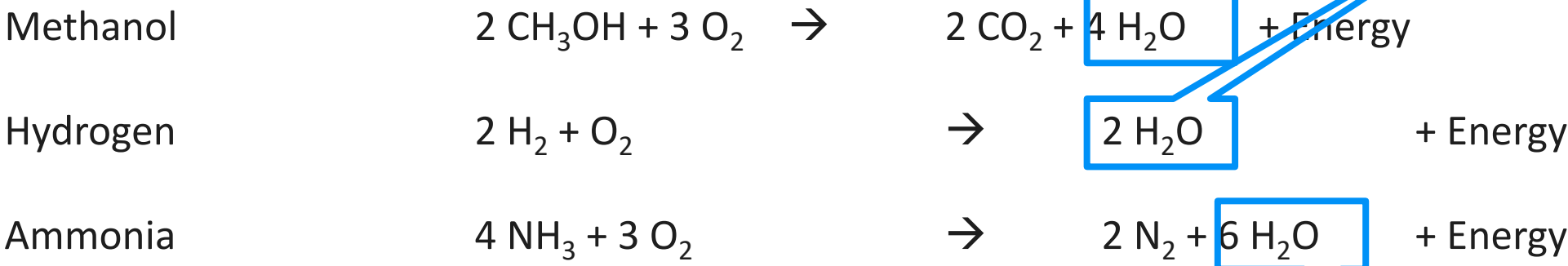
Detection Technologies



Fire and Gas Detection System

Electromagnetic Radiation

➤ Reaction products in combustion processes:



Fire and Gas Detection System

Senses vs. Sensor Technologies

Point Detector



Flammable IR Gas Detector



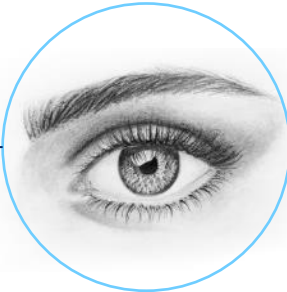
Toxic Electrochemical Gas Detector



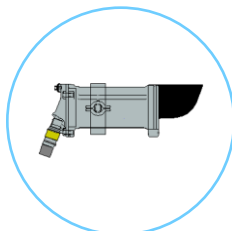
Flammable Catalytic Gas Detector



Area Monitoring / Line of sight



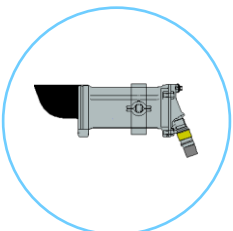
Flame Detector



Open Path Gas Detector



Gas Camera



Ultrasonic Gas Leak Detector



UGLD

Fire and Gas Detection System

Ultrasonic Gas Leak Detection

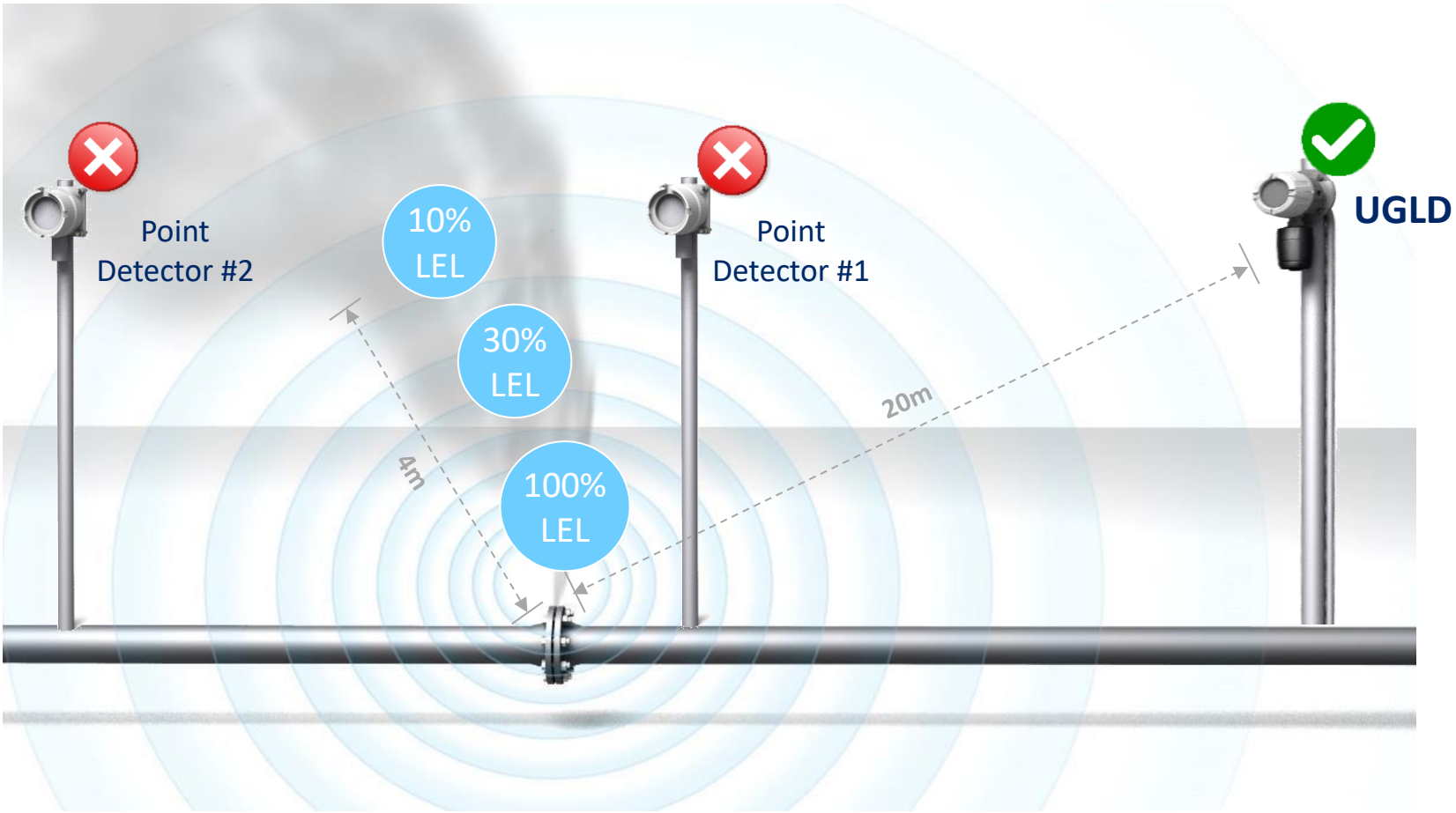
Why Ultrasonic Gas Leak Detection?

Approximately 36% of major gas releases and 69% of significant gas releases are undetected by gas detectors.

Source: Health and Safety Executive (HSE) Offshore Hydrocarbons Release database (2001 – 2008) – (McGillivray and Hare, 2008)

Fire and Gas Detection System

Ultrasonic Gas Leak Detection



Acoustic sensors are robust against wind and gas dilution and offer advantages that cannot be covered by other technologies

- ✓ Alarm
- ✗ No Alarm

A photograph of two firefighters in green helmets and red gear. One firefighter is holding a handheld device, possibly a scanner or communication unit, in a smoky environment. The device has a screen and several buttons. The background is filled with smoke and fire, suggesting a fire scene.

04 Recommendation

Portable Gas Detection Recommendation

Disclaimer: Please check for local rules, regulations and (mining) approvals, they might require different special certification or equipment or require different parameters (such like alarm levels or LEL factors).

Portable Gas Detection

Recommendation for scenarios with H₂



Dräger X-am 2800

Personal Air Monitoring and Protection

- CatEx SR sensor calibrated for Methane with equal sensitivity to H₂
- Dräger Sensor XXS O₂ PR (Poison resistant)
- Dräger Sensor XXS CO LC (cross sensitivity with H₂)

Portable Gas Detection

Recommendation for scenarios with H₂

Dräger X-am 5800



Personal Air Monitoring and Protection

- CatEx SR sensor calibrated for Methane with equal sensitivity to H₂
- Dräger Sensor XXS O₂ PR
- Dräger Sensor XXS H₂ HC (does not show Methane, but 0-4 Vol% H₂)
- Dräger Sensor XXS CO H₂ cp (H₂ compensated; shows less than 15 ppm CO with 1000 ppm H₂ present)

Outlook 2026

- Dräger Sensor IR Ex or IR Ex/CO₂ available (no cross sensitivity to H₂)

Portable Gas Detection

Recommendation for scenarios with H₂



Dräger X-am 8000

Personal Air Monitoring, Leak Detection and Clearance Measurements

- Dräger Dual Sensor IR Ex/CO₂ (no cross sensitivity to H₂)
- Dräger Sensor CatEx H₂ 100 calibrated for H₂ (measures 0-4 Vol% or 0-40.000 ppm (no cross sensitivity to CH₄))
- Dräger Sensor XXS CO H₂ cp cp (H₂ compensated; shows less than 15 ppm CO with 1000 ppm H₂ present)
- Dräger Sensor XXS O₂ / H₂S LC
- Dräger Sensor XXS NO₂
- (DrägerSensor XXS H₂ HC: 0 - 40,000 ppm / 0 - 100 % LEL)

